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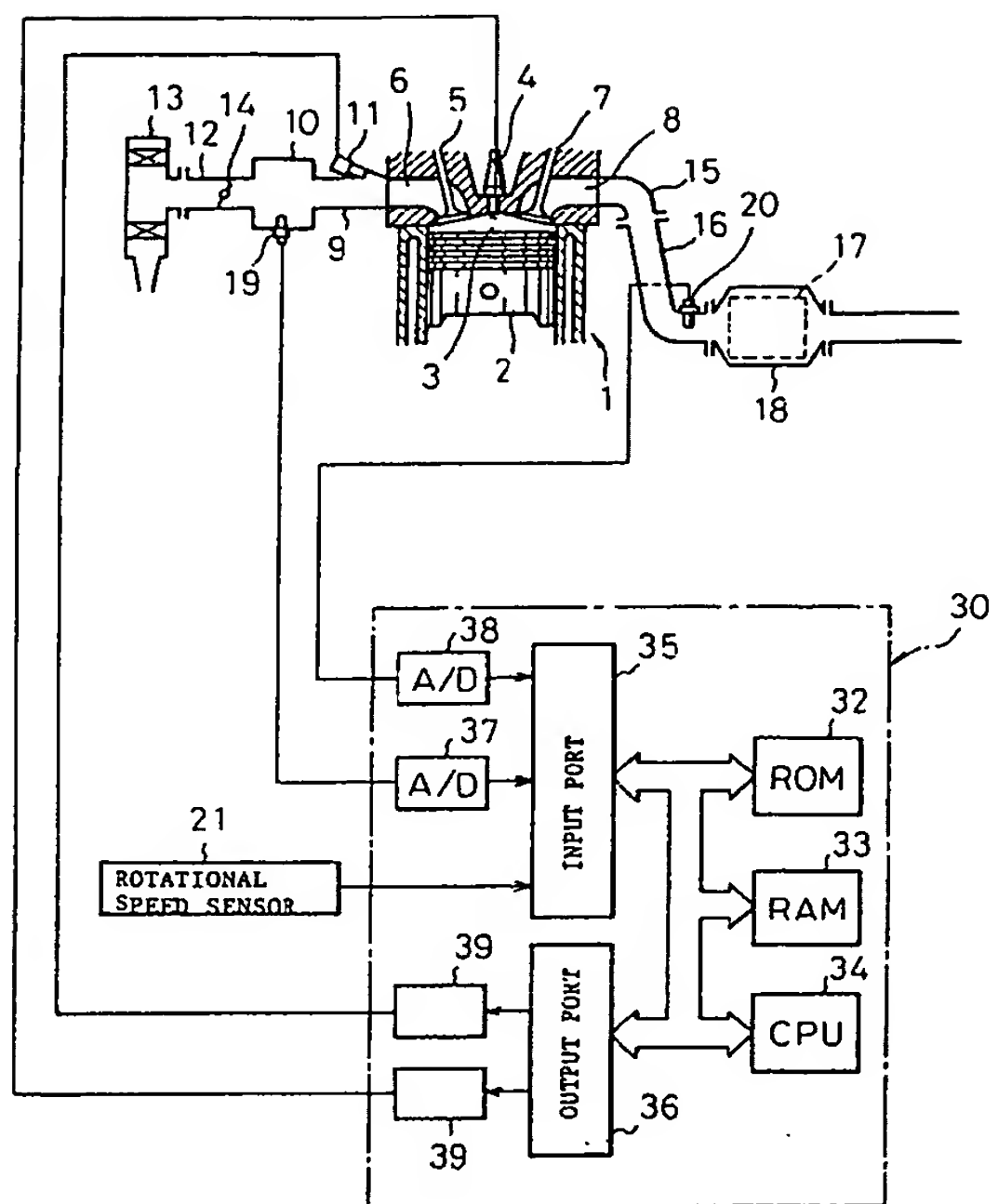
(54) **EXHAUST EMISSION CONTROL SYSTEM FOR INTERNAL COMBUSTION ENGINE.**

(57) An NO_x absorbing agent (17) is disposed inside an exhaust gas passage of a combustion engine. This NO_x absorbing agent (17) absorbs NO_x when the air/fuel ratio of exhaust gas flowing thereinto is lean, and discharges NO_x which it has absorbed when the air/fuel ratio of exhaust gas flowing

thereinto becomes rich. When the air/fuel ratio of exhaust gas flowing into the NO_x absorbing agent (17) is to be made rich so as to discharge NO_x from the NO_x absorbing agent, a higher NO_x absorbing agent temperature makes the degree of richness higher and rich time shorter.

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Fig. 1



TECHNICAL FIELD

The present invention relates to an exhaust purification device of an internal combustion engine.

BACKGROUND ART

A diesel engine in which an engine exhaust passage is branched to a pair of exhaust branch passages for purifying NO_x in the diesel engine, a switching valve is disposed at the branched portion of these exhaust branch passages, the switching valve is switched each time a predetermined time passes to alternately guide the exhaust gas to one of the exhaust branch passages, and a catalyst which can oxidize and absorb the NO_x is disposed in each of the exhaust branch passages is well known (refer to Japanese Unexamined Patent Publication No. 62-106826). In this diesel engine, the NO_x in the exhaust gas introduced into one exhaust branch passage is oxidized and absorbed by the catalyst disposed in that exhaust branch passage. During this time, the inflow of the exhaust gas to the other exhaust branch passage is stopped and, at the same time, a gaseous reducing agent is fed into this exhaust branch passage. The NO_x accumulated in the catalyst disposed in this exhaust branch passage is reduced by this reducing agent. Subsequently, after the elapse of a predetermined time, the introduction of the exhaust gas to the exhaust branch passage to which the exhaust gas had been introduced heretofore is stopped by the switching function of the switching valve, and the introduction of the exhaust gas to the exhaust branch passage to which the introduction of the exhaust gas had been stopped heretofore is started again. That is, in this diesel engine, seen from the viewpoint of each of the exhaust branch passages, exhaust gas is made to flow for a predetermined time during which the NO_x in the exhaust gas is oxidized and absorbed by the catalyst, then the inflow of exhaust gas is stopped for a predetermined period and a reducing agent is fed, whereby the NO_x accumulated in the catalyst is reduced.

However, in this diesel engine, when the NO_x was to be reduced, there was the problem that the inflow of the exhaust gas to the catalyst was stopped, but that the NO_x was attempted to be reduced while causing exhaust gas to flow into the catalyst. That is, with such a catalyst, the speed of reduction of the NO_x changed depending on the temperature of the catalyst and as the temperature of the catalyst became lower, the speed of reduction of the NO_x became slower. Accordingly, where the temperature of the catalyst is low, if a reducing agent is fed for a predetermined period, when the period of feeding the reducing agent is short, the NO_x

absorbed in the catalyst cannot be sufficiently reduced, so a large amount of NO_x remains in the catalyst and therefore the NO_x absorption capacity ends up being reduced. As a result, there was the problem that the NO_x absorption capacity becomes saturated in a short time after the feeding of the reducing agent is stopped and the NO_x absorption action started and therefore the NO_x is released into the atmosphere.

Further, if a large amount of the reducing agent is fed when the temperature of the catalyst is low, only a small amount of the reducing agent is used for the reduction of the NO_x due to the slow speed of reduction of the NO_x , therefore there was the problem that a large amount of the reducing agent is released in the atmosphere. On the other hand, if the amount of the reducing agent is reduced to solve this problem, then even when the temperature of the catalyst is high and therefore the speed of reduction of NO_x is fast, only part of the NO_x is reduced and therefore a large amount of NO_x remains in the catalyst, so the NO_x absorption capacity ends up reduced. As a result, there is the problem that the NO_x absorption capacity ends up saturated in a short time after the feeding of the reducing agent is stopped and the NO_x absorption action is started and therefore the NO_x is released into the atmosphere.

DISCLOSURE OF THE INVENTION

An object of the present invention is to provide an exhaust purification device which suppresses the release of harmful components into the atmosphere at the time of reducing NO_x .

According to the present invention, there is provided an exhaust purification device of an internal combustion engine which has in an engine exhaust passage a NO_x absorbent which absorbs the NO_x when the air-fuel ratio of the inflowing exhaust gas is lean and which releases the absorbed NO_x when the air-fuel ratio of the inflowing exhaust gas becomes rich and which is provided with a temperature detecting means for detecting a temperature representing the temperature of the NO_x absorbent and a NO_x release controlling means for making the air-fuel ratio of the exhaust gas flowing into the NO_x absorbent rich when the NO_x is to be released from the NO_x absorbent and at that time increasing the degree of richness or shortening the time the ratio is made rich the higher the temperature representing the temperature of the NO_x absorbent.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is an overall view of an internal combustion engine; Fig. 2 is a diagram showing a map

of a basic fuel injection time; Fig. 3 is a diagram showing a correction coefficient K; Fig. 4 is a graph schematically showing the concentration of unburnt HC, CO, and oxygen in the exhaust gas discharged from the engine; Fig. 5 is a diagram for explaining an absorption and release function of the NO_x; Fig. 6 is a diagram showing the amount of NO_x discharged from the engine; Fig. 7 is a graph showing the capacity of absorption of NO_x of the NO_x absorbent; Fig. 8 is a graph showing the characteristics of release of NO_x; Fig. 9 is a graph showing the change in the correction coefficient K; Fig. 10 is a graph showing C₁, C₂, α, and β; Fig. 11 is a diagram showing a map of the exhaust gas temperature T; Fig. 12 and Fig. 13 are flow charts showing a time interruption routine; Fig. 14 is a flow chart for calculating the fuel injection time TAU; and Fig. 15 is an overall view of an internal combustion engine showing another embodiment.

BEST MODE FOR CARRYING OUT THE INVENTION

Figure 1 shows a case where the present invention is applied to a gasoline engine.

Referring to Fig. 1, 1 denotes an engine body; 2, a piston; 3, a combustion chamber; 4, a spark plug; 5, an intake valve; 6, an intake port; 7, an exhaust valve; and 8, an exhaust port, respectively. The intake port 6 is connected to a surge tank 10 via a corresponding branch pipe 9, and a fuel injector 11 injecting the fuel toward the interior of the intake port 6 is attached to each branch pipe 9. The surge tank 10 is connected to an air cleaner 13 via an intake duct 12, and a throttle valve 14 is disposed in the intake duct 12. On the other hand, the exhaust port 8 is connected via an exhaust manifold 15 and an exhaust pipe 16 to a casing 18 including a NO_x absorbent 17.

An electronic control unit 30 comprises a digital computer and is provided with a ROM (read only memory) 32, a RAM (random access memory) 33, a CPU (microprocessor) 34, an input port 35, and an output port 36, which are interconnected by a bidirectional bus 31. In the surge tank 10 is mounted a pressure sensor 19 for generating an output voltage proportional to the absolute pressure in the surge tank 10. The output voltage of this pressure sensor 19 is input through an AD converter 37 to the input port 35. Further, a temperature sensor 20 which generates an output voltage proportional to the temperature of the exhaust gas is attached in the exhaust pipe 16 upstream of the casing 18. The output voltage of this temperature sensor 20 is input through an AD converter 38 to the input port 35. Further, the input port 35 is connected to it a rotational speed sensor 21 for generating an output pulse expressing the engine

rotational speed. On the other hand, the output port 36 is connected through a corresponding drive circuit 39 to a spark plug 4 and fuel injector 11.

In the internal combustion engine shown in Fig. 1, the fuel injection time TAU is calculated based on for example the following equation.

$$\text{TAU} = \text{TP} \cdot \text{K}$$

where, TP is a basic fuel injection time, and K is a correction coefficient. The basic fuel injection time TP shows the fuel injection time necessary for bringing the air-fuel ratio of an air-fuel mixture fed into the engine cylinder to the stoichiometric air-fuel ratio. This basic fuel injection time TP is found in advance by experiments and is stored in advance in the ROM 32 in the form of a map as shown in Fig. 2 as the function of the absolute pressure PM in the surge tank 10 and the engine rotational speed N. The correction coefficient K is a coefficient for controlling the air-fuel ratio of the air-fuel mixture fed into the engine cylinder, and if K = 1.0, the air-fuel mixture fed into the engine cylinder becomes the stoichiometric air-fuel ratio. Contrary to this, when K becomes smaller than 1.0, the air-fuel ratio of the air-fuel mixture fed into the engine cylinder becomes larger than the stoichiometric air-fuel ratio, that is, becomes lean, and when K becomes larger than 1.0, the air-fuel ratio of the air-fuel mixture fed into the engine cylinder becomes smaller than the stoichiometric air-fuel ratio, that is, becomes rich.

The value of this correction coefficient K is predetermined in relation to the absolute pressure PM in the surge tank 10 and the engine rotational speed N. Figure 3 shows an embodiment of the value of the correction coefficient K. In the embodiment shown in Fig. 3, in the region where the absolute pressure PM in the surge tank 10 is relatively low, that is, in the engine low and medium load operation region, the value of the correction coefficient K is made a value smaller than 1.0, therefore at this time the air-fuel ratio of the air-fuel mixture fed into the engine cylinder is made lean. On the other hand, in the region where the absolute pressure PM in the surge tank 10 is relatively high, that is, in the engine high load operation region, the value of the correction coefficient is made 1.0. Accordingly, at this time, the air-fuel ratio of the air-fuel mixture fed into the engine cylinder is made the stoichiometric air-fuel ratio. Further, in the region where the absolute pressure PM in the surge tank 10 becomes the highest, that is, in the engine full load operation region, the value of the correction coefficient is made a value larger than 1.0. Therefore, at this time, the air-fuel ratio of the air-fuel mixture fed into the engine cylinder is made rich. An internal combustion engine is usually op-

erated most frequently with a low and medium load and therefore for the majority of the period of operation a lean air-fuel mixture is burned.

Figure 4 schematically shows the concentration of representative components in the exhaust gas discharged from the combustion chamber 3. As seen from Fig. 4, the concentration of the unburnt HC and CO in the exhaust gas discharged from the combustion chamber 3 is increased as the air-fuel ratio of the air-fuel mixture fed into the combustion chamber 3 becomes richer, and the concentration of the oxygen O_2 in the exhaust gas discharged from the combustion chamber 3 is increased as the air-fuel ratio of the air-fuel mixture fed into the combustion chamber 3 becomes leaner.

The NO_x absorbent 17 contained in the casing 18 uses, for example, alumina as a carrier. On this carrier, at least one substance selected from alkali metals, for example, potassium K, sodium Na, lithium Li, and cesium Cs; alkali earths, for example, barium Ba and calcium Ca; and rare earths, for example, lanthanum La and yttrium Y and a precious metal such as platinum Pt are carried. When referring to the ratio between the air and fuel (hydrocarbons) fed into the intake passage of the engine and the exhaust passage upstream of the NO_x absorbent 17 as the air-fuel ratio of the inflowing exhaust gas flowing into the NO_x absorbent 17, this NO_x absorbent 17 performs the absorption and releasing function of NO_x by absorbing the NO_x when the air-fuel ratio of the inflowing exhaust gas is lean, while releasing the absorbed NO_x when the concentration of oxygen in the inflowing exhaust gas falls. Note that, where the fuel (hydrocarbons) or air is not fed into the exhaust passage upstream of the NO_x absorbent 17, the air-fuel ratio of the inflowing exhaust gas coincides with the air-fuel ratio of the air-fuel mixture fed into the combustion chamber 3, and accordingly in this case, the NO_x absorbent 17 absorbs the NO_x when the air-fuel ratio of the air-fuel mixture fed into the combustion chamber 3 is lean and releases the absorbed NO_x when the concentration of oxygen in the air-fuel mixture fed into the combustion chamber 3 is lowered.

When the above-mentioned NO_x absorbent 17 is disposed in the exhaust passage of the engine, this NO_x absorbent 17 actually performs the absorption and releasing function of NO_x , but there are areas of the exact mechanism of this absorption and releasing function which are not clear. However, it can be considered that this absorption and releasing function is conducted by the mechanism as shown in Fig. 5. This mechanism will be explained by using as an example a case where platinum Pt and barium Ba are carried on the carrier, but a similar mechanism is obtained even if another precious metal, alkali metal, alkali earth, or

rare earth is used.

Namely, when the inflowing exhaust gas becomes considerably lean, the concentration of oxygen in the inflowing exhaust gas is greatly increased. As shown in Fig. 5(A), the oxygen O_2 is deposited on the surface of the platinum Pt in the form of O_2^- or O^{2-} . On the other hand, the NO in the inflowing exhaust gas reacts with the O_2^- or O^{2-} on the surface of the platinum Pt and becomes NO_2 ($2NO + O_2 \rightarrow 2NO_2$). Subsequently, a part of the produced NO_2 is oxidized on the platinum Pt and absorbed into the absorbent. While bonding with the barium oxide BaO, it is diffused in the absorbent in the form of nitric acid ions NO_3^- as shown in Fig. 5(A). In this way, NO_x is absorbed into the NO_x absorbent 17.

So long as the oxygen concentration in the inflowing exhaust gas is high, the NO_2 is produced on the surface of the platinum Pt, and so long as the NO_x absorption ability of the absorbent is not saturated, the NO_2 is absorbed into the absorbent and nitric acid ions NO_3^- are produced. Contrary to this, when the oxygen concentration in the inflowing exhaust gas is lowered and the production of NO_2 is lowered, the reaction proceeds in an inverse direction ($NO_3^- \rightarrow NO_2$), and thus nitric acid ions NO_3^- in the absorbent are released in the form of NO_2 from the absorbent. Namely, when the oxygen concentration in the inflowing exhaust gas is lowered, the NO_x is released from the NO_x absorbent 17. As shown in Fig. 4, when the degree of leanness of the inflowing exhaust gas becomes low, the oxygen concentration in the inflowing exhaust gas is lowered, and accordingly when the degree of leanness of the inflowing exhaust gas is lowered, the NO_x is released from the NO_x absorbent 17 even if the air-fuel ratio of the inflowing exhaust gas is lean.

On the other hand, at this time, when the air-fuel ratio of the air-fuel mixture fed into the combustion chamber 3 is made rich and the air-fuel ratio of the inflowing exhaust gas becomes rich, as shown in Fig. 4, a large amount of unburnt HC and CO is discharged from the engine, and these unburnt HC and CO react with the oxygen O_2^- or O^{2-} on the platinum Pt and are oxidized. Also, when the air-fuel ratio of the inflowing exhaust gas becomes rich, the oxygen concentration in the inflowing exhaust gas is extremely lowered, and therefore the NO_2 is discharged from the absorbent. This NO_2 reacts with the unburnt HC and CO as shown in Fig. 5(B) and is reduced. In this way, when the NO_2 no longer exists on the surface of the platinum Pt, the NO_2 is successively released from the absorbent. Accordingly, when the air-fuel ratio of the inflowing exhaust gas is made rich, the NO_x is released from the NO_x absorbent 19 in a short time.

Namely, when the air-fuel ratio of the inflowing exhaust gas is made rich, first of all, the unburnt HC and CO immediately react with the O_2^- or O^{2-} on the platinum Pt and are oxidized, and subsequently if the unburnt HC and CO still remain even though the O_2^- or O^{2-} on the platinum Pt is consumed, the NO_x released from the absorbent and the NO_x discharged from the engine are reduced by these unburnt HC and CO. Accordingly, when the air-fuel ratio of the inflowing exhaust gas is made rich, the NO_x absorbed in the NO_x absorbent 17 is released in a short time and in addition this released NO_x is reduced, and therefore the discharge of NO_x into the atmosphere can be blocked. Also, since the NO_x absorbent 17 has the function of a reduction catalyst, even if the air-fuel ratio of the inflowing exhaust gas is made the stoichiometric air-fuel ratio, the NO_x released from the NO_x absorbent 17 can be reduced. However, where the air-fuel ratio of the inflowing exhaust gas is made the stoichiometric air-fuel ratio, the NO_x is released merely gradually from the NO_x absorbent 17, and therefore a slightly long time is required for releasing all NO_x absorbed in the NO_x absorbent 17.

When the degree of leanness of the inflowing exhaust gas is lowered as mentioned before, even if the air-fuel ratio of the inflowing exhaust gas is lean, the NO_x is released from the NO_x absorbent 17. Accordingly, so as to release the NO_x from the NO_x absorbent 17, it is satisfactory if the concentration of oxygen in the inflowing exhaust gas is lowered. Note, even if the NO_x is released from the NO_x absorbent 17, when the air-fuel ratio of the inflowing exhaust gas is lean, the NO_x is not reduced in the NO_x absorbent 17, and accordingly, in this case, it is necessary to provide a catalyst which can reduce the NO_x downstream of the NO_x absorbent 17 or supply a reducing agent downstream of the NO_x absorbent 17. Of course, it is also possible to reduce the NO_x downstream of the NO_x absorbent 17 in this way, but it is rather preferable that the NO_x be reduced in the NO_x absorbent 17. Accordingly, in the embodiment according to the present invention, when the NO_x should be released from the NO_x absorbent 17, the air-fuel ratio of the inflowing exhaust gas is made rich, whereby the NO_x released from the NO_x absorbent 17 is reduced in the NO_x absorbent 17.

However, in the embodiment according to the present invention, as mentioned above, during full load operation, the air-fuel mixture fed into the engine cylinder 3 is made rich and during high load operation, the air-fuel mixture is made the stoichiometric air-fuel ratio, so during the full load operation and the high load operation, the NO_x is released from the NO_x absorbent 17. However, if the frequency of this full load operation or high

load operation is small, then even if the NO_x is released from the NO_x absorbent 17 only during full load operation and high load operation, the absorption capacity of the NO_x by the NO_x absorbent 17 will end up becoming saturated during the time when a lean air-fuel mixture is burnt and therefore it will end up becoming impossible for the NO_x absorbent 17 to absorb the NO_x . Accordingly, in the embodiment according to the present invention, when a lean air-fuel mixture continues to be burnt, the air-fuel mixture fed into the combustion chamber 3 is cyclically made rich and during this time the NO_x is released from the NO_x absorbent 17.

In this case, however, if the cycle at which the air-fuel mixture fed into the engine cylinder 3 is made rich is long, then the NO_x absorbing capacity of the NO_x absorbent 17 will end up becoming saturated during the time the lean air-fuel mixture is being burnt and therefore the NO_x can no longer be absorbed in the NO_x absorbent 17, so there will be the problem that NO_x will end up being released into the atmosphere. As opposed to this, even if an engine operating state where a large amount of NO_x is discharged from the engine continues, if the cycle at which the air-fuel mixture is made lean is shortened so that the NO_x is released from the NO_x absorbent 17 before the NO_x absorbing capacity of the NO_x absorbent 17 becomes saturated, then this time the problem will arise of an increase of the amount of fuel consumption.

Therefore, in the present invention, the amount of NO_x which is absorbed in the NO_x absorbent 17 is found and the air-fuel mixture is made rich when the amount of the NO_x absorbed in the NO_x absorbent 17 exceeds a predetermined allowable value. If the air-fuel mixture is made rich when the amount of the NO_x absorbed in the NO_x absorbent 17 exceeds a predetermined allowable value, then the NO_x absorbing capacity of the NO_x absorbent 17 will never become saturated, so the NO_x will no longer be released into the atmosphere and, further, the frequency at which the air-fuel mixture is made rich can be reduced as well, so it is possible to suppress an increase in the amount of the fuel consumption.

However, when finding the amount of NO_x being absorbed in the NO_x absorbent 17, it is difficult to directly find the amount of NO_x being absorbed in the NO_x absorbent 17. Therefore, in the embodiment according to the present invention, the amount of the NO_x absorbed in the NO_x absorbent 17 is estimated from the amount of NO_x in the exhaust gas discharged from the engine. That is, the higher the rotational speed N of the engine, the larger the amount of exhaust gas discharged per unit time from the engine, so as the engine rotational speed N becomes higher, the amount of NO_x

discharged from the engine per unit time increases. Further, the higher the engine load, that is, the higher the absolute pressure PM in the surge tank 10, the greater the amount of the exhaust gas discharged from the combustion chambers 3 and further the higher the combustion temperature, so the higher the engine load, that is, the higher the absolute pressure PM in the surge tank 10, the greater the amount of NO_x discharged from the engine per unit time.

Figure 6(A) shows the relationship between the amount of the NO_x discharged from the engine per unit time, the absolute pressure PM in the surge tank 10, and the engine rotational speed N as found by experiments. In Fig. 6(A), the curves show the identical amounts of NO_x. As shown in Fig. 6(A), the amount of NO_x discharged from the engine per unit time becomes larger the higher the absolute pressure PM in the surge tank 10 and becomes larger the higher the engine rotational speed N. Note that the amount of NO_x shown in Fig. 6(A) is stored in the ROM 32 in advance in the form of a map as shown in Fig. 6(B).

On the other hand, Fig. 7 shows the relationship between the absorption capacity NO_xCAP which can be absorbed by the NO_x absorbent 17 and the temperature T of the exhaust gas, which represents the temperature of the NO_x absorbent 17. If the temperature of the NO_x absorbent 17 becomes lower, that is, the temperature T of the exhaust gas becomes lower, the oxidation action of the NO_x ($2\text{NO} + \text{O}_2 \rightarrow 2\text{NO}_2$) is weakened, and therefore the NO_x absorption capacity NO_xCAP is lowered. Further, if the temperature of the NO_x absorbent 17 becomes higher, that is, the temperature T of the exhaust gas becomes higher, the NO_x absorbed in the NO_x absorbent 17 is decomposed and naturally released, so the NO_x absorption capacity NO_xCAP is lowered. Accordingly, the NO_x absorption capacity NO_xCAP becomes larger between about 300 °C to about 500 °C.

On the other hand, Fig. 8 shows the results of experiments on the amount of NO_x released from the NO_x absorbent 17 when switching the air-fuel ratio of the exhaust gas flowing into the NO_x absorbent 17 from lean to rich. Note that the solid line in Fig. 8 shows the state when the temperature of the NO_x absorbent 17, that is, the temperature T of the exhaust gas, is high, while the broken line shows when the temperature of the NO_x absorbent 17, that is, the temperature T of the exhaust gas, is low. The rate of decomposition of the NO_x in the NO_x absorbent 17 becomes faster the higher the temperature of the NO_x absorbent 17. Therefore, when the temperature of the NO_x absorbent 17 is high, as shown by the solid line in Fig. 8, that is, when the temperature T of the exhaust gas is high, a large amount of NO_x is released from the NO_x

absorbent 17 in a short time, while when the temperature of the NO_x absorbent 17, that is, the temperature T of the exhaust gas, is low, as shown by the broken line in Fig. 8, a small amount of NO_x is continually released from the NO_x absorbent 17 over a long period. That is, the higher the temperature T of the exhaust gas, the greater the amount of NO_x released from the NO_x absorbent 17 per unit time and the shorter the release time of the NO_x.

When the amount of the unburnt HC and CO discharged from the engine, however, is smaller than the amount which can reduce the total NO_x released from the NO_x absorbent 17, part of the NO_x is released into the atmosphere without being reduced, while when the amount of unburnt HC and CO discharged from the engine is greater than the amount able to reduce the total NO_x released from the NO_x absorbent 17, the excess unburnt HC and CO are released into the atmosphere. Accordingly, to prevent the NO_x and the unburnt HC and CO from being released into the atmosphere, it is necessary to discharge exactly the amount of the unburnt HC and CO from the engine needed to reduce the NO_x released from the NO_x absorbent 17. Toward this end, it becomes necessary to increase the amount of the unburnt HC and CO in accordance with the curve shown in Fig. 8.

As mentioned earlier, however, the amount of the unburnt HC and CO discharged from the engine is proportional to the degree of richness of the air-fuel mixture fed into the combustion chamber 3. Therefore, in the present invention, as shown in Fig. 9, the value of the correction coefficient k with respect to the basic fuel injection time TP, that is, the degree of richness of the air-fuel mixture, is made to change in accordance with a pattern as close as possible to the pattern of change of the concentration of NO_x shown in Fig. 8. Note that here, the correction coefficient k has the relationship $K = 1 + k$ with the above-mentioned correction coefficient K and therefore when $k = 0$, the air-fuel mixture becomes the stoichiometric air-fuel ratio while when $k > 0$, the air-fuel mixture becomes rich. Therefore, it is understood from Fig. 9 that the higher the temperature of the NO_x absorbent 17, the higher the degree of richness and the shorter the time the ratio is made rich.

As shown by the solid line in Fig. 9, when the NO_x is to be released from the NO_x absorbent 17, the correction coefficient k is made to rise by α increments with each passing of the unit time until the time C reaches C₁. Next, when the time C is between C₁ and C₂, the correction coefficient k is held constant, then when the time C exceeds C₂, the correction coefficient k is made to descend in β decrements with each unit time. The values of these α , β , C₁, and C₂ are set so that the pattern of

change of the correction coefficient k becomes as close as possible to the pattern of change of the concentration of NO_x shown by the solid line in Fig. 8.

On the other hand, the pattern of change of the correction coefficient k when the temperature of the NO_x absorbent 17, that is, the temperature T of the exhaust gas, is low, is also set so that it becomes as close as possible to the pattern of change of the concentration of NO_x when the temperature T of the exhaust gas is low, as shown by the broken line in Fig. 8. In this case, to make the pattern of change of the correction coefficient k in Fig. 9 like the broken line, it is understood that it is sufficient to make both α and β smaller and make C_1 and C_2 larger. That is, to make the pattern of change of the correction coefficient k close to the pattern of change of the concentration of NO_x shown in Fig. 8, it is sufficient to make α and β larger and make C_1 and C_2 smaller as the temperature T of the exhaust gas becomes higher, as shown in Fig. 10. Note that the relationship between C_1 , C_2 , α , and β and the temperature T of the exhaust gas shown in Fig. 10 is stored in advance in the ROM 32.

Note that in the embodiment according to the present invention, provision is made of a temperature sensor 25 for detecting the temperature T of the exhaust gas and accordingly the NO_x absorption capacity $\text{NO}_x \text{ CAP}$ shown in Fig. 7 and the α , β , C_1 , and C_2 shown in Fig. 10 are determined based on the temperature T of the exhaust gas detected by this temperature sensor 25. The temperature T of the exhaust gas, however, can be estimated from the absolute pressure PM in the surge tank 10 and the engine rotational speed N . Therefore, instead of providing the temperature sensor 25, it is possible to store the temperature T of the exhaust gas in the ROM 32 in advance in the form of a map as shown in Fig. 11 and determine the NO_x absorption capacity $\text{NO}_x \text{ CAP}$ and α , β , C_1 , and C_2 based on the temperature T of the exhaust gas obtained from this map.

Next, an explanation will be made of the first embodiment of control of the release of NO_x with reference to Fig. 12 to Fig. 14.

Figure 12 and Fig. 13 show a time interruption routine executed by interruption every predetermined time.

Referring to Fig. 12 and Fig. 13, first, at step 100, it is judged if a NO_x release flag showing that the NO_x should be released from the NO_x absorbent 17 is set or not. When the NO_x release flag is not set, the routine proceeds to step 101, where it is judged if the correction coefficient K is smaller than 1.0, that is, if the operating state is one in which the air-fuel mixture should be made lean. When $K < 1.0$, that is, when the operating state is one in which the air-fuel mixture should be made

lean, the routine proceeds to step 102, where the count D is made zero, then the routine proceeds to step 103.

At step 103, the NO_x amount N_{ij} discharged from the engine per unit time is calculated from the map shown in Fig. 6(B) based on the absolute pressure PM in the surge tank 10, detected by the pressure sensor 19, and the engine rotational speed N . Next, at step 104, the NO_x amount N_{ij} is multiplied by the interruption time interval Δt and the product $N_{ij} \cdot \Delta t$ is added to ΣNO_x . The product $N_{ij} \cdot \Delta t$ shows the amount of the NO_x discharged from the engine during the interruption time interval Δt . At this time, the NO_x discharged from the engine is absorbed by the NO_x absorbent 17, so ΣNO_x shows the estimated value of the amount of NO_x absorbed in the NO_x absorbent 17.

Next, at step 105, the NO_x absorption capacity $\text{NO}_x \text{ CAP}$ is calculated from the relationship shown in Fig. 7 based on the temperature T of the exhaust gas detected by the temperature sensor 25. Next, at step 106, it is judged if the estimated value ΣNO_x of the amount of NO_x absorbed in the NO_x absorbent 17 has exceeded the NO_x absorption capacity $\text{NO}_x \text{ CAP}$. When $\Sigma \text{NO}_x \leq \text{NO}_x \text{ CAP}$, the processing cycle is completed. At this time, a lean air-fuel mixture is burned and the NO_x discharged from the engine is absorbed in the NO_x absorbent 17.

On the other hand, if it is judged at step 106 that $\Sigma \text{NO}_x > \text{NO}_x \text{ CAP}$, that is, the NO_x absorption capacity of the NO_x absorbent 17 is saturated, the routine proceeds to step 107, where the NO_x release flag is set. Next, at step 108, C_1 , C_2 , α , and β are calculated from the relation shown in Fig. 10 based on the temperature T of the exhaust gas and the processing cycle is ended. If the NO_x release flag is set, at the next processing cycle, the routine proceeds from step 100 to step 109, where the count C is incremented by one. Next, at step 110, it is judged if the count C is smaller than C_1 . When $C < C_1$, the routine proceeds to step 111, where α is added to the correction coefficient k . Next, the processing cycle is ended. The action of addition of α to the correction coefficient k is performed continuously until $C \geq C_1$. Accordingly, the value of the correction coefficient k during this time continues to increase as shown in Fig. 9.

On the other hand, if it is judged at step 110 that $C \geq C_1$, the routine proceeds to step 112, where it is judged if the count C has become smaller than C_2 . When $C < C_2$, the processing cycle is ended. Therefore, as shown in Fig. 9, the correction coefficient k is held constant until $C \geq C_2$.

Next, at step 112, when it is judged that $C \geq C_2$, the routine proceeds to step 113, where β is subtracted from the correction coefficient k . Next,

at step 113, it is judged if the correction coefficient k has become zero or a negative number. When $k > 0$, the processing cycle is ended. Accordingly, as shown in Fig. 9, the correction coefficient k is reduced until $k \leq 0$. Note that, as mentioned later, if $k > 0$, the air-fuel mixture fed to the combustion chamber 3 is made rich and during this time the degree of richness is changed by the pattern shown in Fig. 9.

On the other hand, if it is judged at step 114 that $K \leq 0$, the routine proceeds to step 115, where the NO_x release flag is reset. Next, at step 116, ΣNO_x is made zero. That is, at this time, it is considered that all of the NO_x which had been absorbed in the NO_x absorbent 17 is released, so the estimated value ΣNO_x of the NO_x absorbed in the NO_x absorbent 17 is made zero. Next, at step 117, the count C and the correction coefficient k are made zero and the processing cycle is ended.

On the other hand, if it is judged at step 101 that $k \geq 1.0$, that is, when the engine operating state is one in which the air-fuel mixture should be made rich or the stoichiometric air-fuel ratio, the routine proceeds to step 118, where the count D is incremented by one. Next, at step 119, it is judged if the count D has become larger than the constant value D_0 . When $D > D_0$, the routine proceeds to step 120, where ΣNO_x is made zero. That is, when the combustion of the rich air-fuel ratio or the stoichiometric air-fuel ratio continues for a certain time, it may be considered that all of the NO_x has been released from the NO_x absorbent 17, so at this time the estimated value ΣNO_x of the amount of NO_x absorbed in the NO_x absorbent 17 is made zero.

Figure 14 shows the routine for calculation of the fuel injection time TAU . This routine is repeatedly executed.

Referring to Fig. 14, first, at step 150, a basic fuel injection time TP is calculated from a map indicated in Fig. 2. Subsequently, at step 151, the correction coefficient K shown in Fig. 3, which is determined in accordance with the operating state of the engine, is calculated. Next, at step 152, it is judged if the NO_x release flag is set or not. When the NO_x release flag is not set, the routine proceeds to step 153, where the correction coefficient K is made K_1 . Next, at step 155, K_1 is multiplied with the basic fuel injection time TP , whereby the fuel injection time TAU is calculated. Accordingly, at this time, the air-fuel mixture which is fed into the combustion chamber 3 is made lean, the stoichiometric air-fuel ratio, or rich in accordance with the operating state of the engine as shown in Fig. 3.

On the other hand, when it is judged at step 152 that the NO_x release flag is set, the routine proceeds to step 154, where K_1 is made the sum (k

+ 1) of correction coefficient k calculated by the routine shown in Fig. 12 and Fig. 13 and 1, then the routine proceeds to step 155. Next, at this time, the air-fuel mixture fed to the combustion chamber 3 is made rich, then the degree of richness is changed by the pattern shown in Fig. 9.

Figure 15 shows another embodiment of the internal combustion engine. In this embodiment, an outlet side of a casing 18 is connected through an exhaust pipe 22 to a catalytic converter 24 housing a three-way catalyst 23. This three-way catalyst 23, as is well known, exhibits a high purification efficiency with respect to CO , HC , and NO_x when the air-fuel ratio is maintained near the stoichiometric air-fuel ratio, but the three-way catalyst 23 also has a high purification efficiency with respect to NO_x even when the air-fuel ratio becomes rich to a certain extent. In the embodiment shown in Fig. 15, a three-way catalyst 23 is provided downstream of the NO_x absorbent 17 so as to remove the NO_x using this characteristic.

That is, as is mentioned above, if the air-fuel mixture fed into the engine cylinder is made rich to release the NO_x from the NO_x absorbent 17, the NO_x absorbed in the NO_x absorbent 17 is rapidly released from the NO_x absorbent 17. At this time, the NO_x is reduced during its release, but there is a possibility that all of the NO_x will not be reduced. If the three-way catalyst 23 is disposed downstream of the NO_x absorbent 17, however, the NO_x which could not be reduced at the time of the release of the NO_x is reduced by the three-way catalyst 23. Accordingly, by disposing the three-way catalyst 23 downstream of the NO_x absorbent 17, it becomes possible to improve considerably the purification performance of the NO_x .

In the embodiments discussed up to here, use was made, as the NO_x absorbent, of a NO_x absorbent 17 comprised of at least one of an alkali metal, alkali earth, and rare earth and a precious metal carried on alumina. Instead of using such a NO_x absorbent 17, however, it is also possible to use a complex oxide of an alkali earth and copper, that is, a NO_x absorbent of the Ba-Cu-O system. As such a complex oxide of an alkali earth and copper, use may be made for example of $\text{MnO}_2 \cdot \text{BaCuO}_2$. In this case, it is also possible to add platinum Pt or cerium Ce . In a NO_x absorbent of the $\text{MnO}_2 \cdot \text{BaCuO}_2$ system, the copper Cu performs the same catalytic function as the platinum Pt in the NO_x absorbent 17 spoken of up to now. When the air-fuel ratio is lean, the NO_x is oxidized by the copper ($2\text{NO} + \text{O}_2 \rightarrow 2\text{NO}_2$) and dispersed in the absorbent in the form of nitric acid ions NO_3^- .

On the other hand, if the air-fuel ratio is rich, similarly, NO_x is released from the absorbent. This NO_x is reduced by the catalytic action of the cop-

per Cu. The NO_x reducing ability of copper Cu, however, is weaker than the NO_x reducing ability of platinum Pt and therefore when using an absorbent of the Ba-Cu-O system, the amount of NO_x which is not reduced at the time of release of the NO_x becomes somewhat greater than with the NO_x absorbent 17 discussed up to now. Therefore, when using an absorbent of the Ba-Cu-O system, as shown in Fig. 21, it is preferable to dispose a three-way catalyst 23 downstream of the absorbent.

LIST OF REFERENCE NUMERALS

- 3 combustion chamber
- 5 intake valve
- 7 exhaust valve
- 16 exhaust pipe
- 17 NO_x absorbent
- 25 temperature sensor

Claims

1. An exhaust purification device of an internal combustion engine which has in an engine exhaust passage a NO_x absorbent which absorbs the NO_x when the air-fuel ratio of the inflowing exhaust gas is lean and which releases the absorbed NO_x when the air-fuel ratio of the inflowing exhaust gas becomes rich and which is provided with a temperature detecting means for detecting a temperature representing the temperature of the NO_x absorbent and a NO_x release controlling means for making the air-fuel ratio of the exhaust gas flowing into the NO_x absorbent rich when the NO_x is to be released from the NO_x absorbent and at that time increasing the degree of richness or shortening the time the ratio is made rich the higher the temperature representing the temperature of the NO_x absorbent.
2. An exhaust purification device of an internal combustion engine according to claim 1, wherein said NO_x release controlling means increases the degree of richness and makes the time of making the ratio richer shorter the higher the temperature representing the temperature of the NO_x absorbent.
3. An exhaust purification device of an internal combustion engine according to claim 2, wherein said NO_x release controlling means changes the degree of richness by a pattern of change resembling the pattern of change of the amount of NO_x released from the NO_x absorbent when releasing NO_x from the NO_x absorbent.
4. An exhaust purification device of an internal combustion engine according to claim 3, wherein said NO_x release controlling means increases the degree of richness to a predetermined degree of richness by a predetermined rate of rise and then reduces the degree of richness by a predetermined rate of reduction when releasing NO_x from the NO_x absorbent and increases the rate of rise, the predetermined degree of richness, and the rate of reduction the higher the temperature representing the temperature of the NO_x absorbent.
5. An exhaust purification device of an internal combustion engine according to claim 1, wherein the temperature representing the NO_x absorbent is the temperature of the exhaust gas flowing into the NO_x absorbent.
6. An exhaust purification device of an internal combustion engine according to claim 5, wherein provision is made of a memory in which is previously stored the temperature of the exhaust gas flowing into the NO_x absorbent as a function of the engine load and the engine rotational speed and wherein the temperature representing the temperature of the NO_x absorbent is stored in the memory.
7. An exhaust purification device of an internal combustion engine according to claim 1, wherein said NO_x release controlling means makes the air-fuel ratio of the air-fuel mixture fed into the combustion chamber rich when releasing NO_x from the NO_x absorbent.
8. An exhaust purification device of an internal combustion engine according to claim 1, wherein NO_x estimating means is provided for estimating the amount of NO_x absorbed in said NO_x absorbent and wherein the NO_x release controlling means makes the air-fuel ratio of the exhaust gas flowing into the NO_x absorbent rich to release the NO_x from the NO_x absorbent when the amount of NO_x estimated to be absorbed in the NO_x absorbent by the NO_x estimating means exceeds a predetermined allowance.
9. An exhaust purification device of an internal combustion engine according to claim 8, wherein said NO_x estimating means estimates the amount of the NO_x absorbed in the NO_x absorbent on the basis of the amount of NO_x discharged from the combustion chamber to the engine exhaust passage.

10. An exhaust purification device of an internal combustion engine according to claim 9, wherein said NO_x estimating means is comprised of a NO_x calculating means for calculating the amount of NO_x discharged per unit time from the engine to the engine exhaust passage in accordance with the engine load and the engine rotational speed and an cumulative adding means for cumulatively adding the amounts of NO_x calculated by the NO_x calculating means.
11. An exhaust purification device of an internal combustion engine according to claim 10, wherein said NO_x calculating means is provided with a memory in which is previously stored the amount of NO_x discharged per unit time from the engine to the engine exhaust passage as a function of the engine load and the engine rotational speed and wherein the cumulative adding means cumulatively adds the amounts of NO_x stored in the memory and determined from the engine load and the engine rotational speed.
12. An exhaust purification device of an internal combustion engine according to claim 10, wherein provision is made of a throttle valve disposed in the engine intake passage for controlling the engine load and the vacuum inside the engine intake passage downstream of the throttle valve is used as a value representing the engine load.
13. An exhaust purification device of an internal combustion engine according to claim 8, wherein said allowance is the maximum NO_x absorption capacity of the NO_x absorbent.
14. An exhaust purification device of an internal combustion engine according to claim 13, wherein said allowance is a function of the temperature representing the temperature of the NO_x absorbent.
15. An exhaust purification device of an internal combustion engine according to claim 8, wherein said NO_x estimating means makes the amount of NO_x estimated to be absorbed in the NO_x absorbent zero when the air-fuel ratio of the exhaust gas flowing into the NO_x absorbent is made rich due to the amount of NO_x estimated by the NO_x estimating means exceeding the allowance.
16. An exhaust purification device of an internal combustion engine according to claim 8, wherein air-fuel ratio control means is provided

for making the air-fuel ratio of the air-fuel mixture fed to the combustion chamber rich in accordance with the operating state of the engine regardless of whether the amount of NO_x estimated by the NO_x estimating means exceeds the allowance.

17. An exhaust purification device of an internal combustion engine according to claim 16, wherein said air-fuel ratio control means makes the air-fuel ratio of the air-fuel mixture fed into the combustion chamber rich when the engine load is higher than a predetermined load.
18. An exhaust purification device of an internal combustion engine according to claim 17, wherein said NO_x estimating means makes the amount of NO_x estimated as being absorbed in the NO_x absorbent zero when the air-fuel ratio of the air-fuel mixture is made rich for more than a predetermined time.
19. An exhaust purification device of an internal combustion engine according to claim 1, wherein the NO_x absorbent includes at least one substance selected from alkali metals such as potassium, sodium, lithium, and cesium, alkali earths such as barium and calcium, and rare earths such as lanthanum and yttrium and platinum.
20. An exhaust purification device of an internal combustion engine according to claim 1, wherein the NO_x absorbent is comprised of a compound oxide of barium and copper.
21. An exhaust purification device of an internal combustion engine according to claim 1, wherein a three-way catalyst is disposed in the engine exhaust passage downstream of the NO_x absorbent.

Fig.1

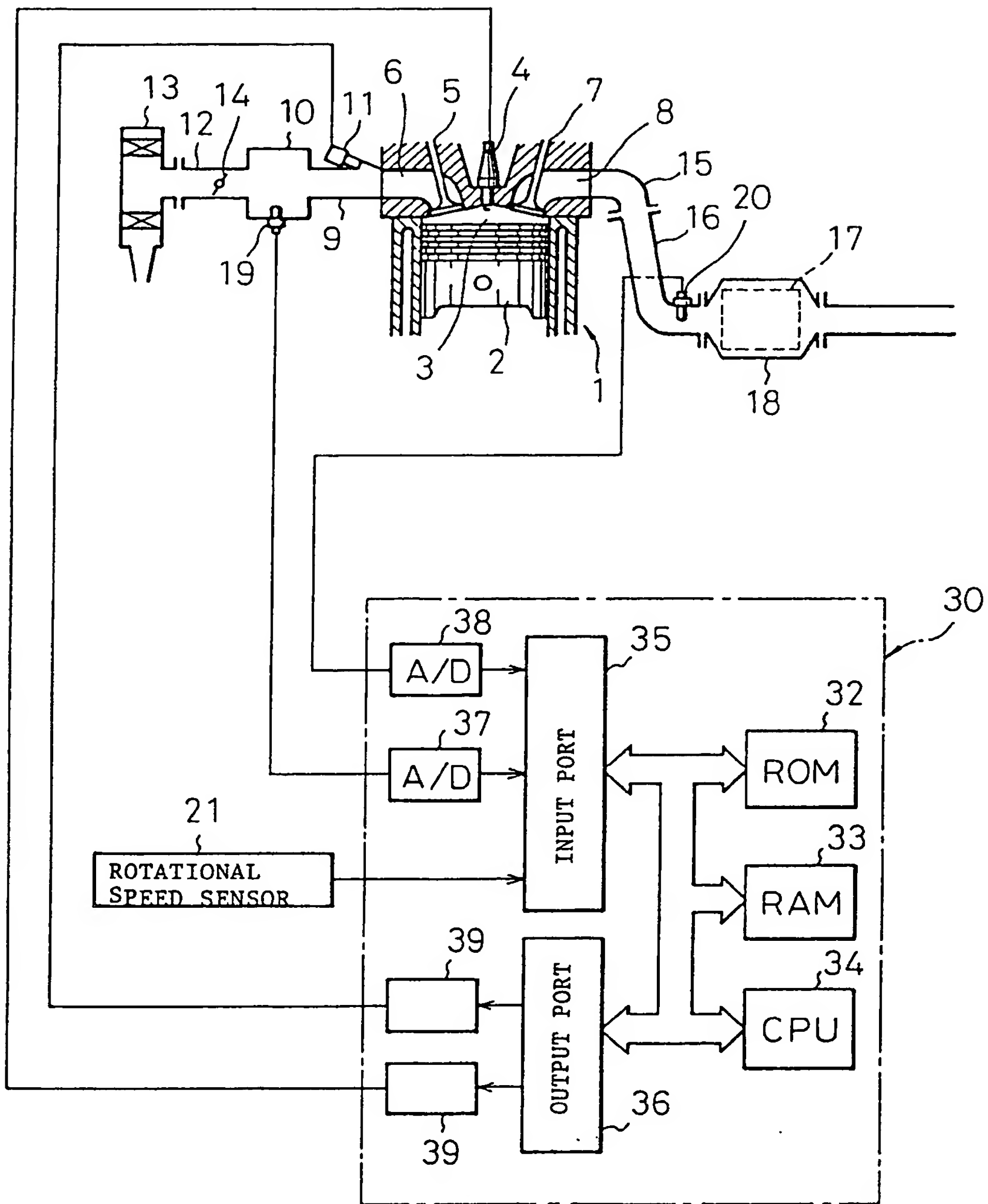


Fig.2

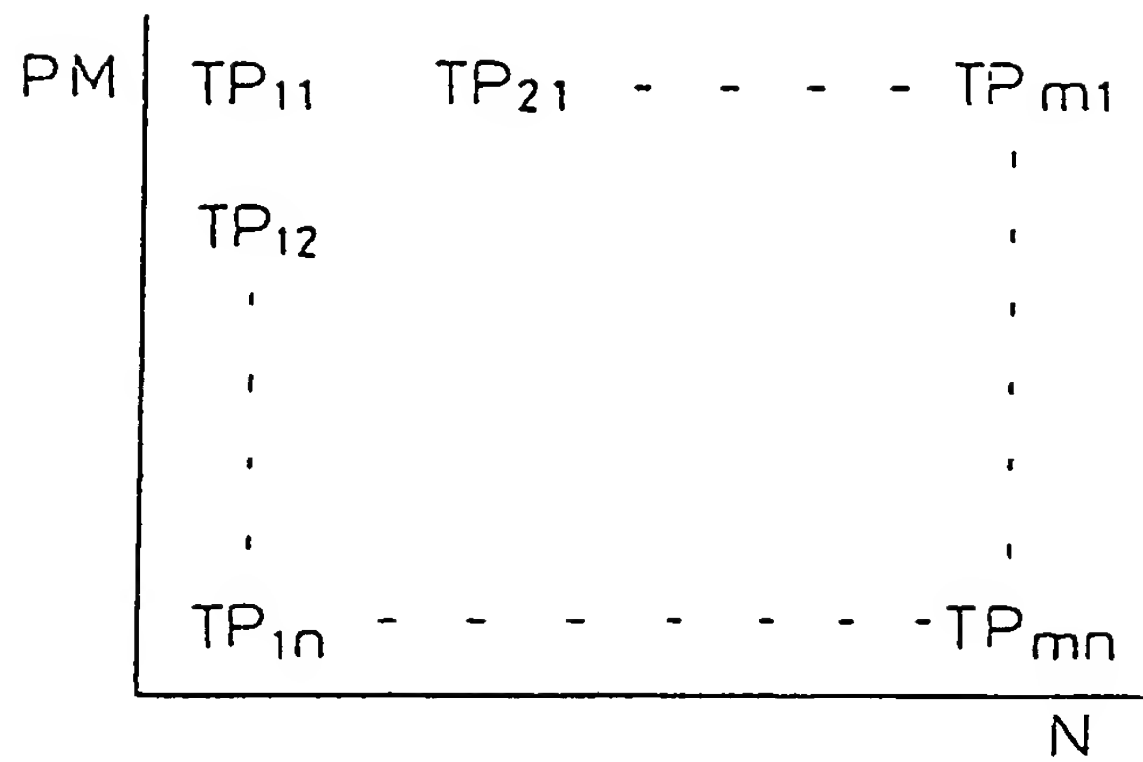


Fig.3

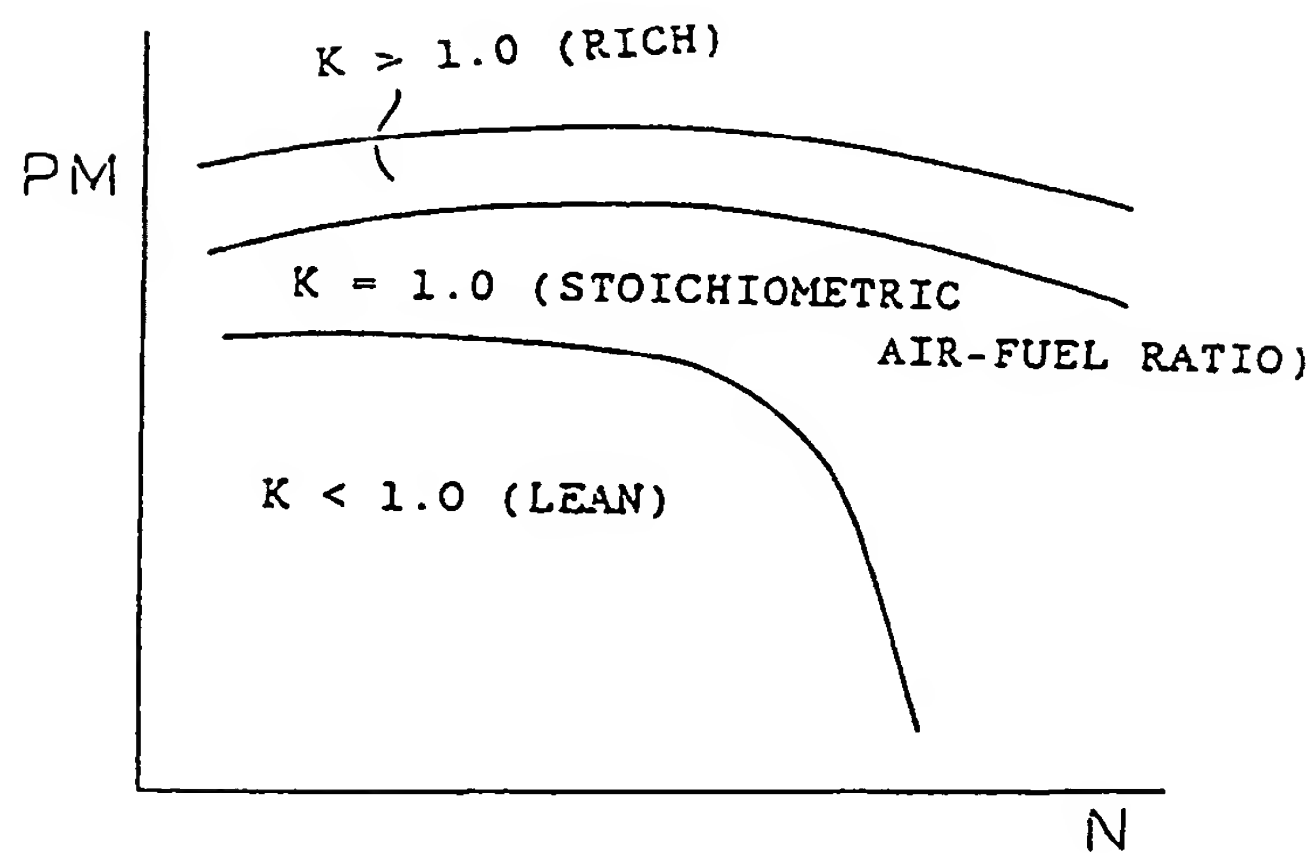


Fig.4

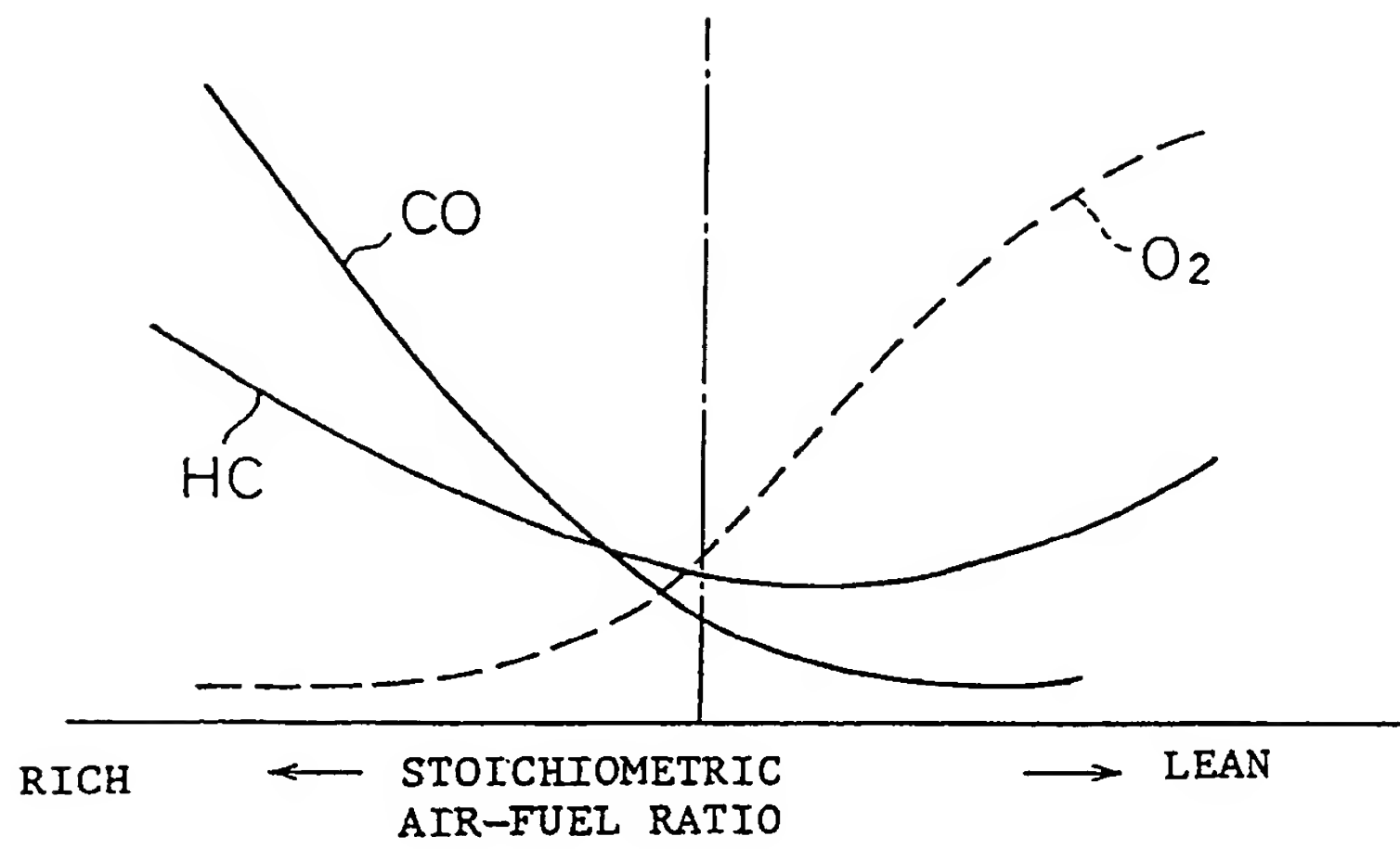


Fig.5(A)

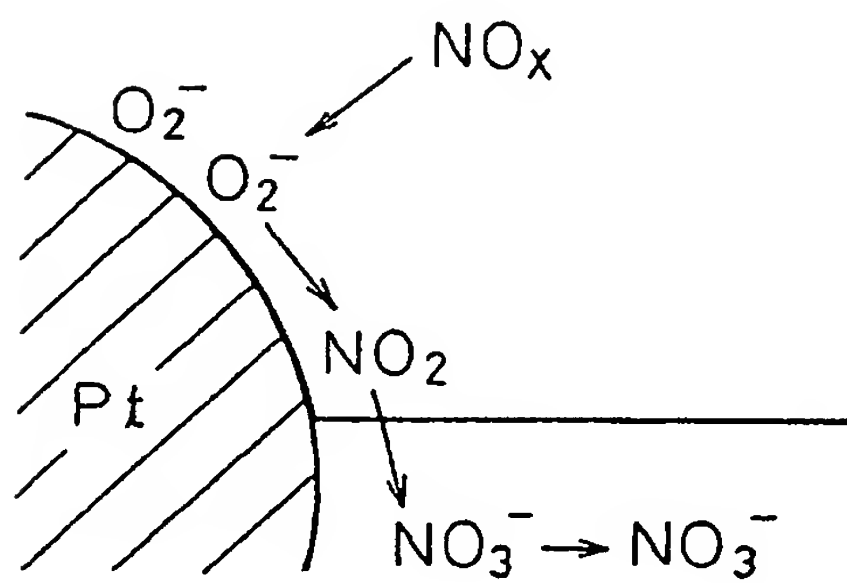


Fig.5(B)

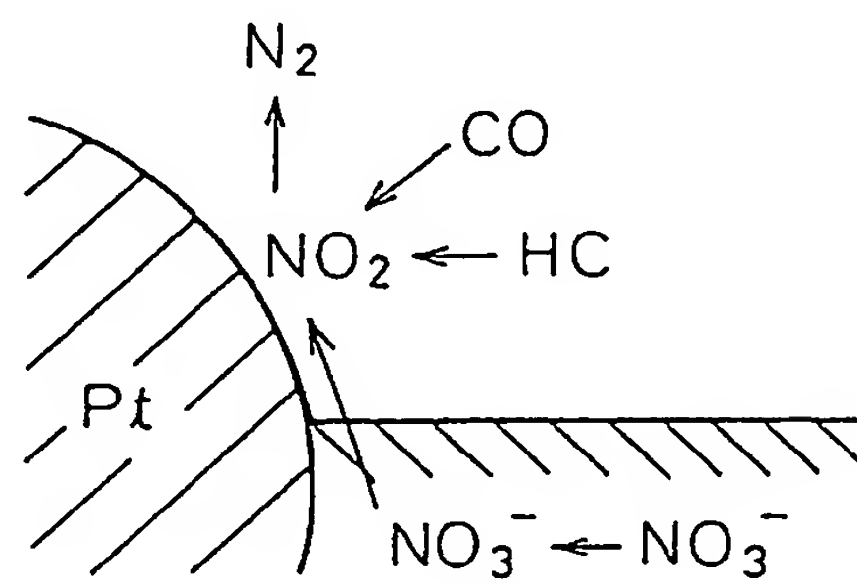


Fig.6(A)

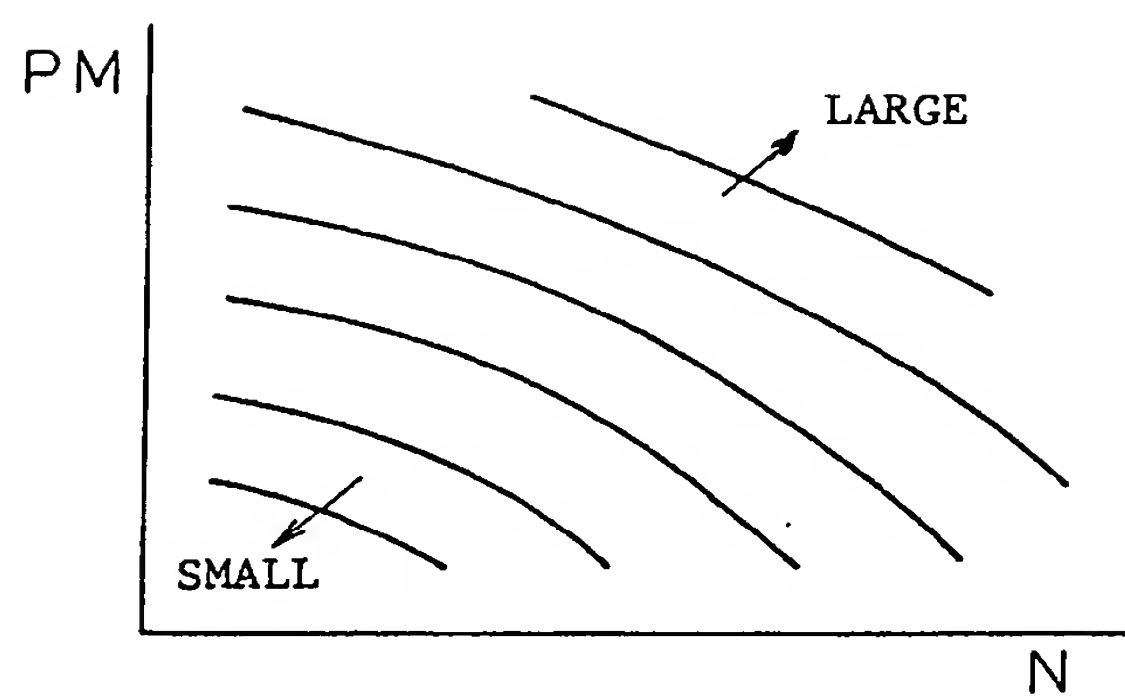


Fig.6(B)

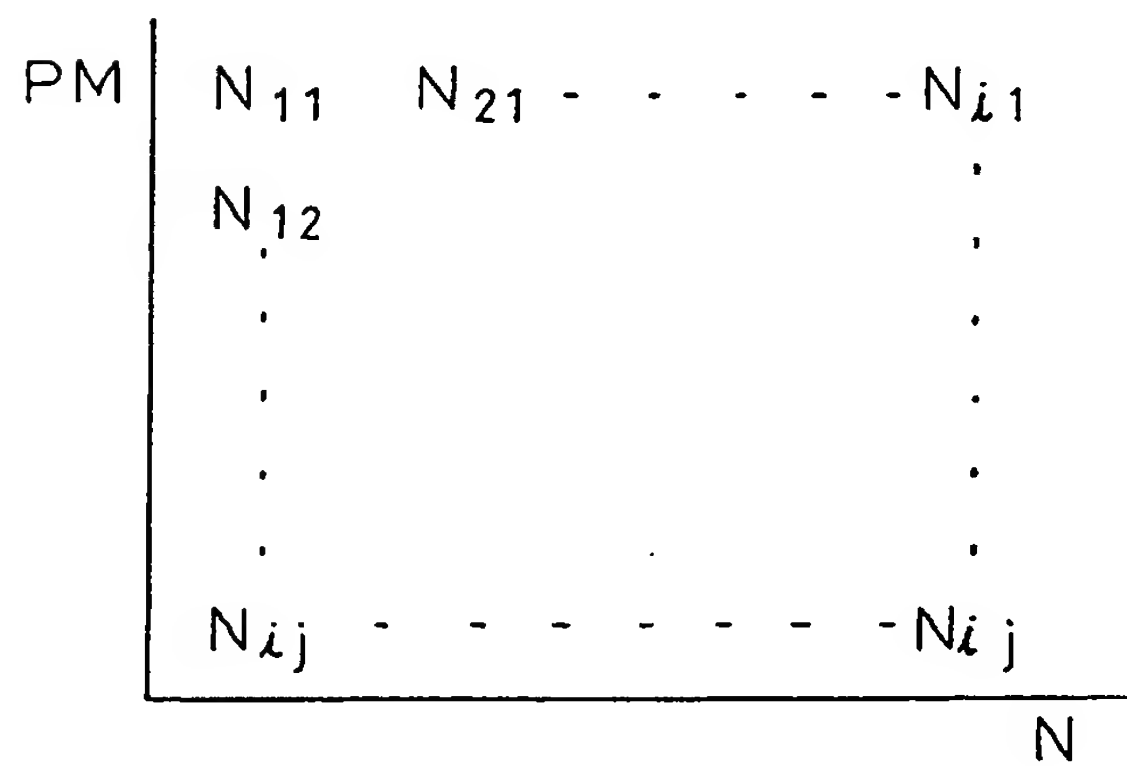


Fig.7

NO_x ABSORPTION
CAPACITY
NO_x CAP

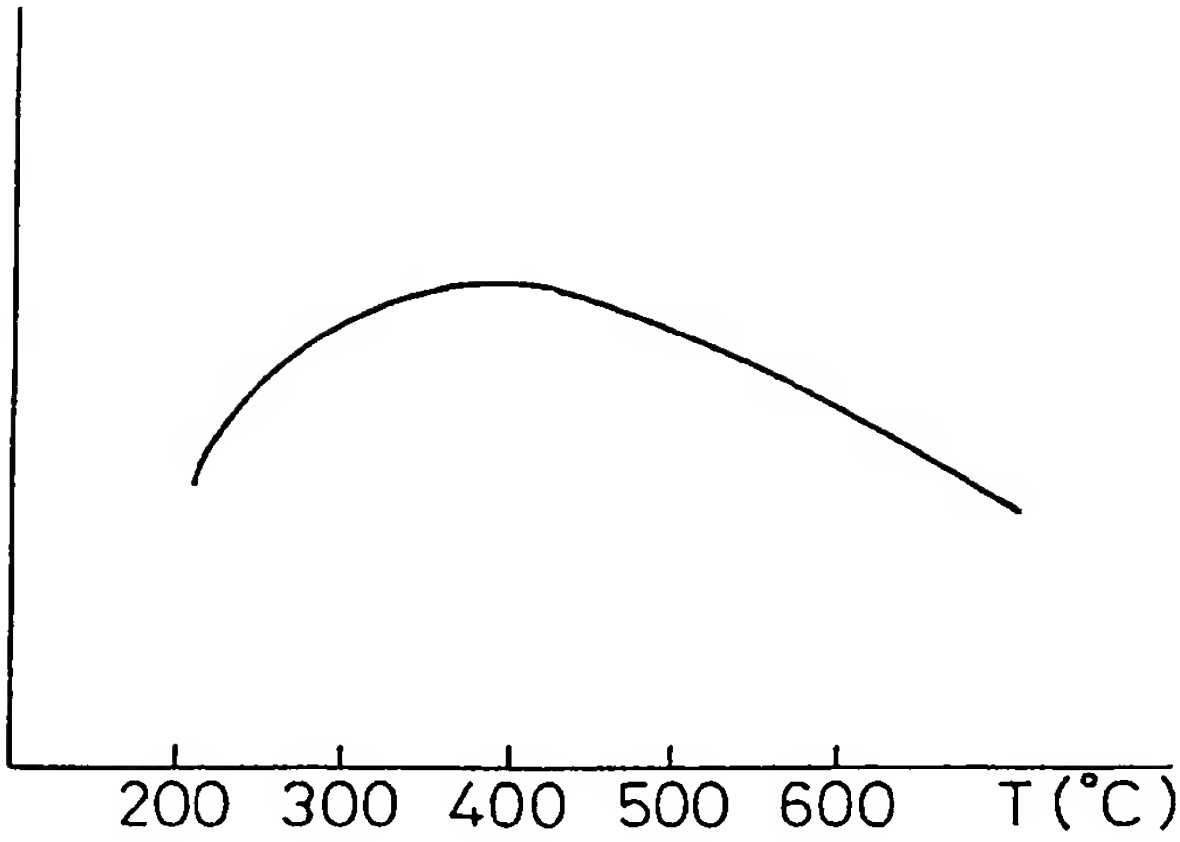


Fig.8

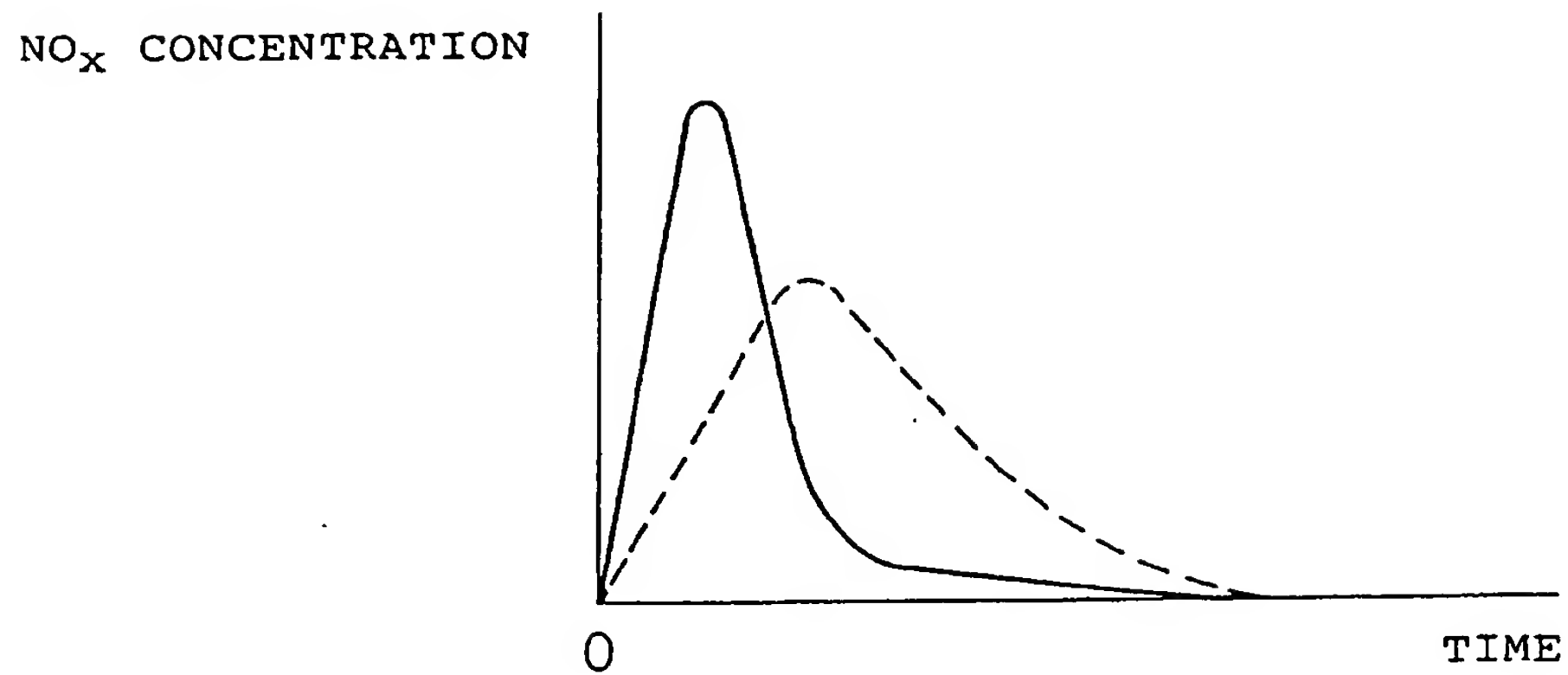


Fig.9

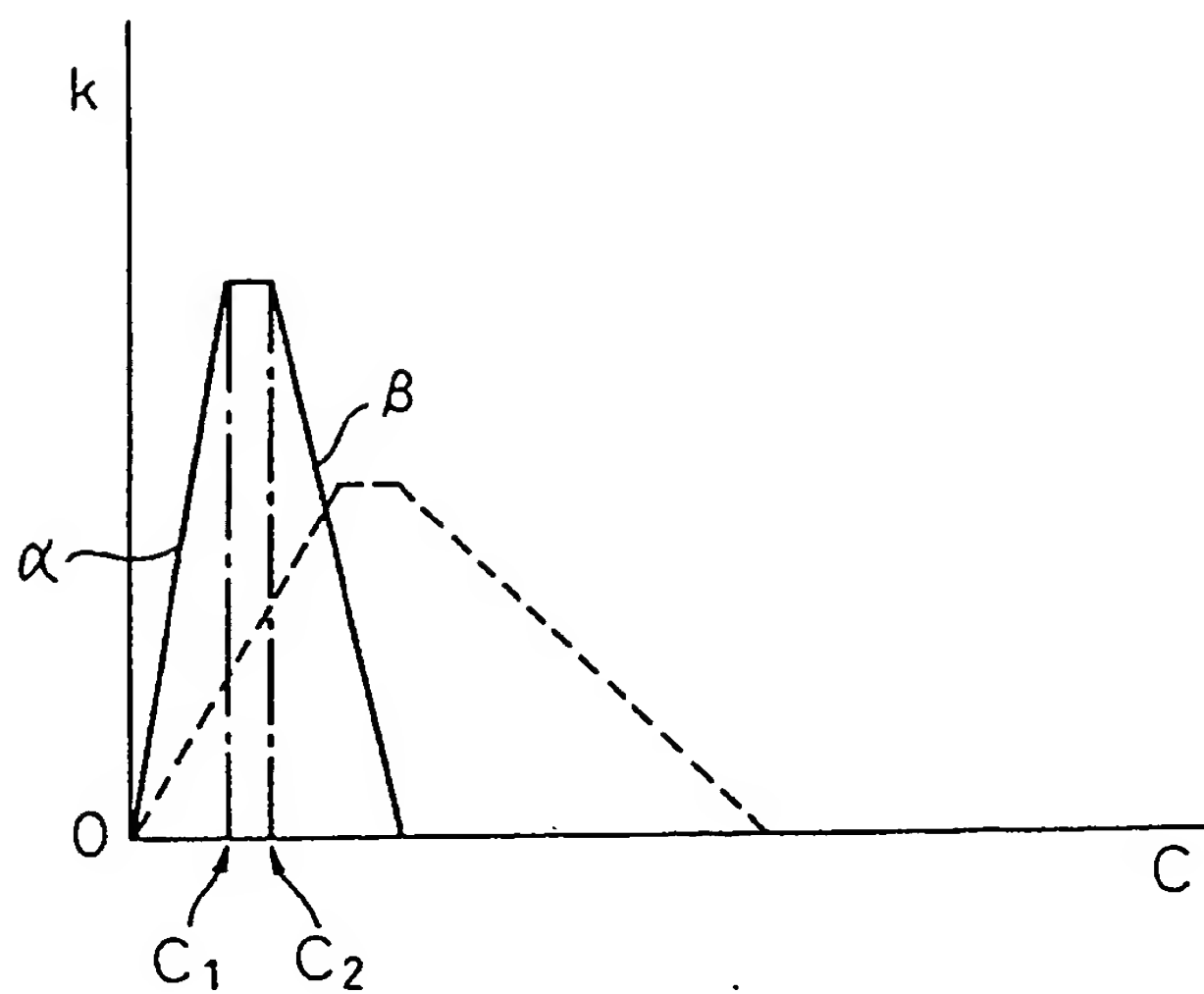


Fig.10

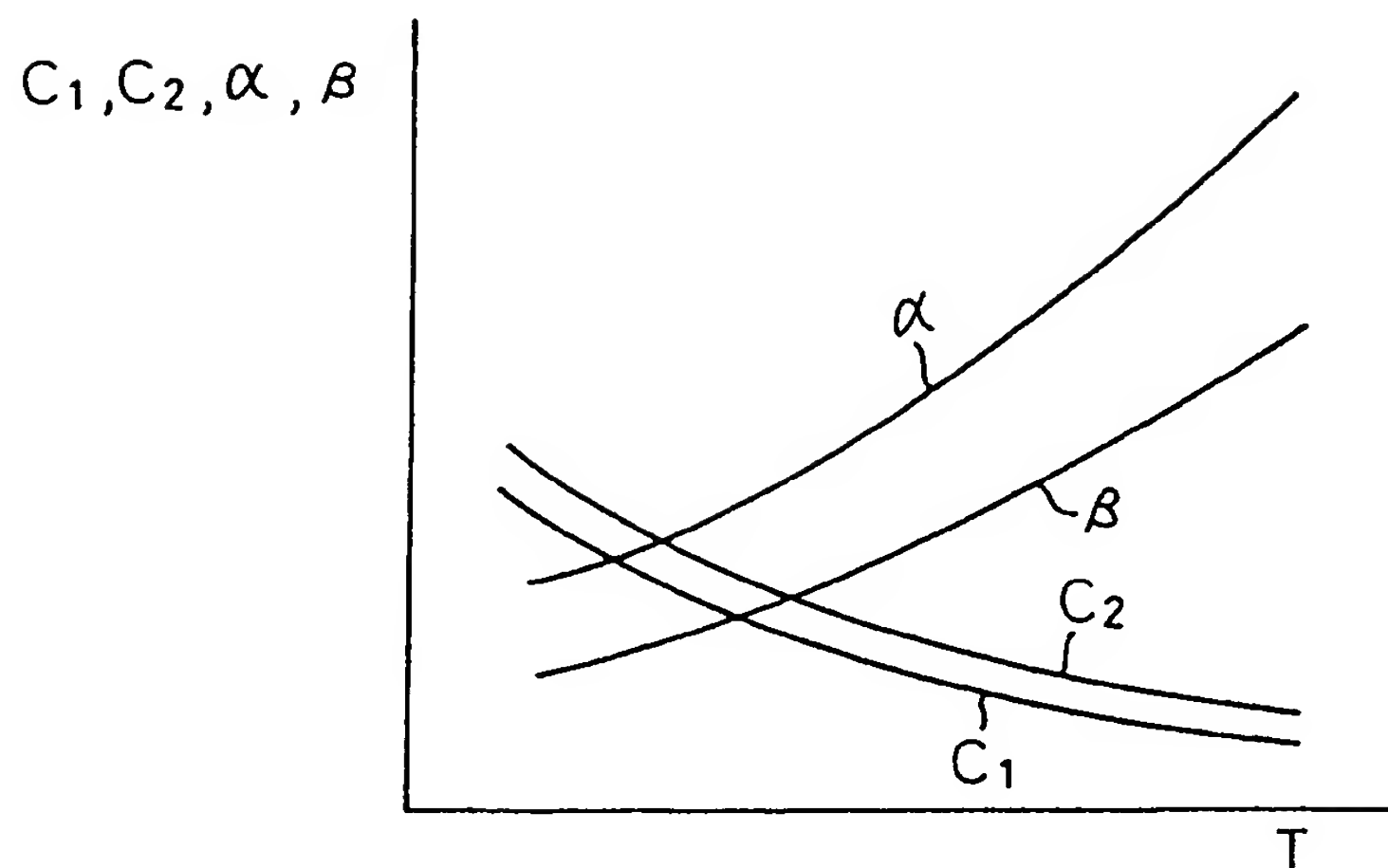


Fig.11

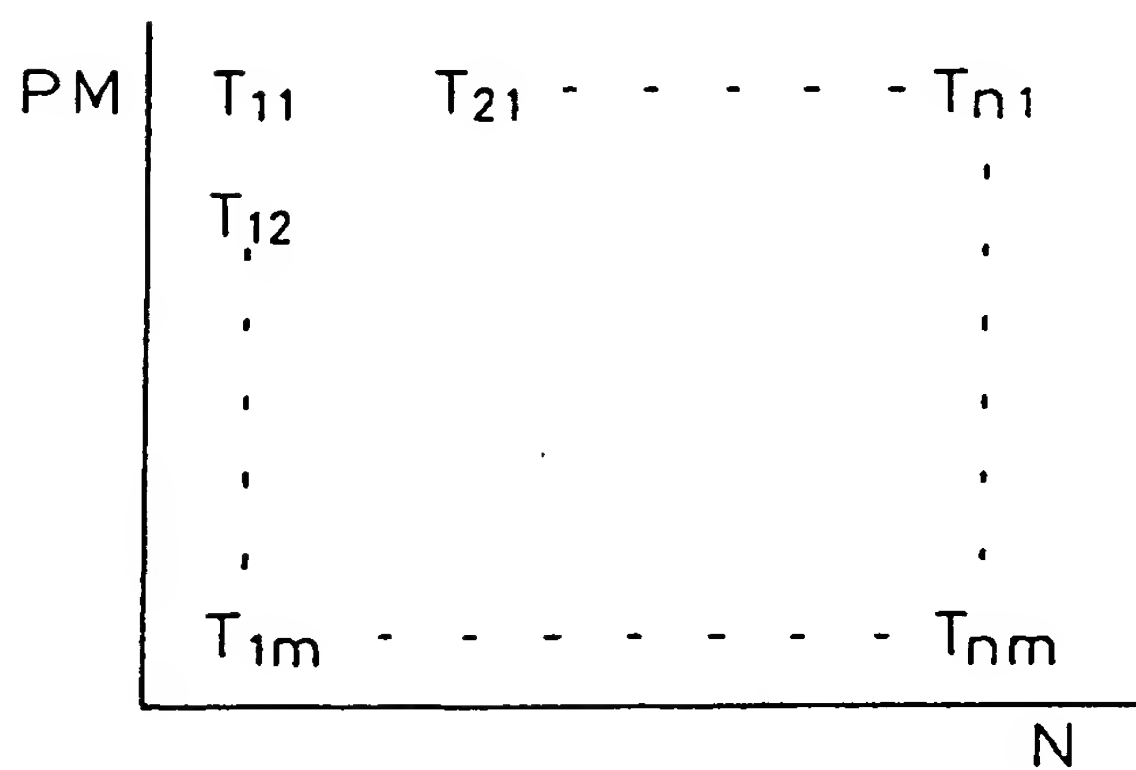


Fig. 12

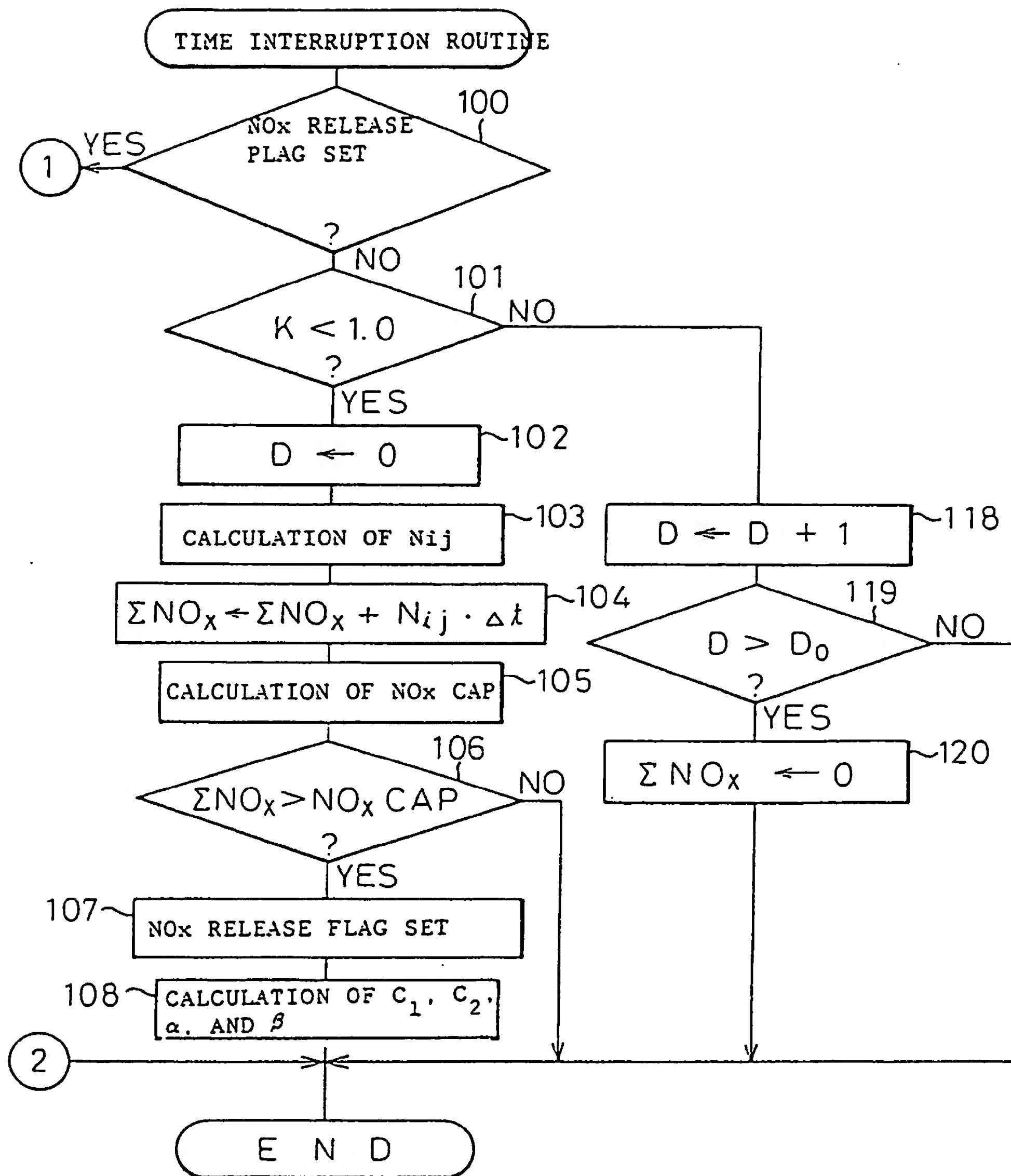


Fig.13

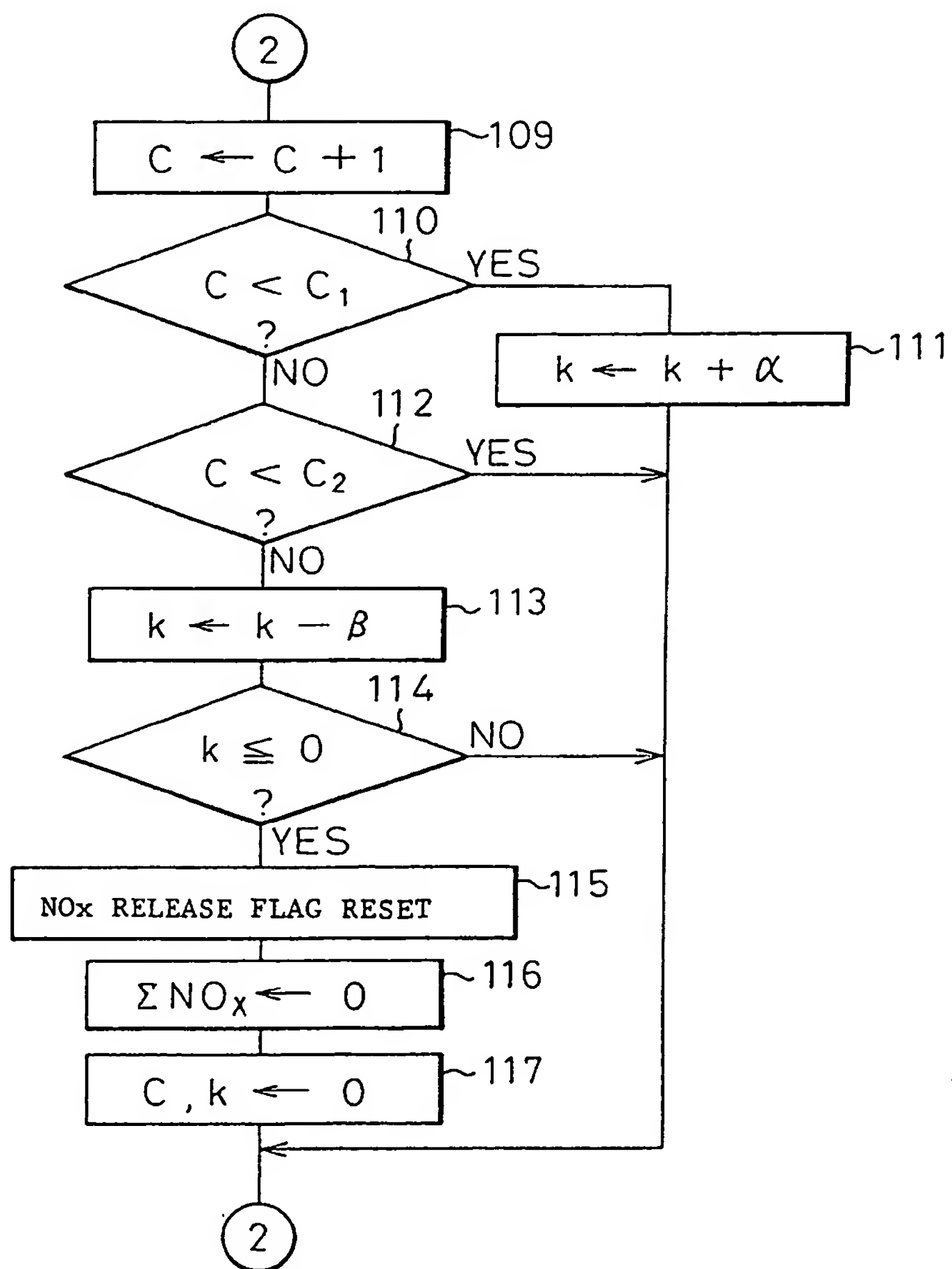


Fig.14

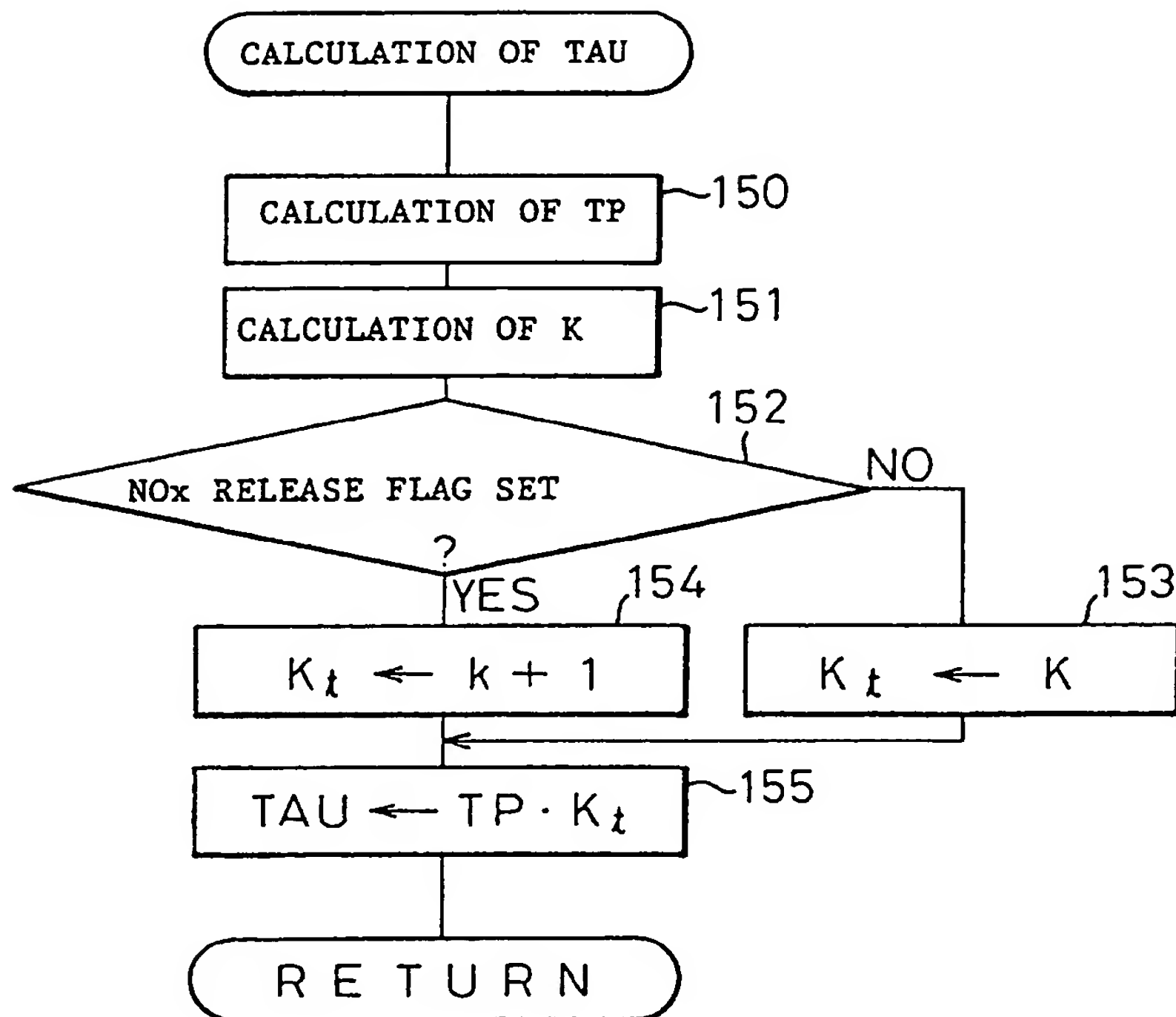
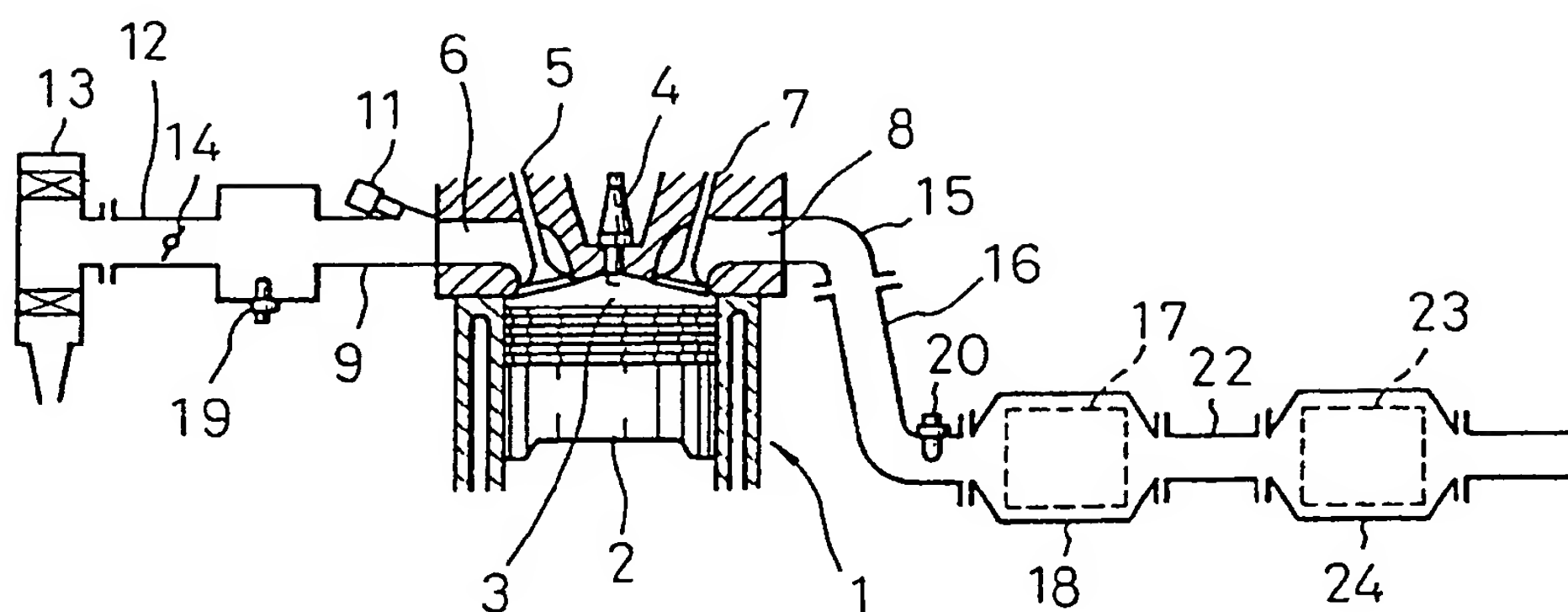


Fig.15



INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP93/00777

A. CLASSIFICATION OF SUBJECT MATTER

Int. Cl⁵ F01N3/28, F01N3/24, B01D53/34, B01J23/58, F02D41/02

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

Int. Cl⁵ F01N3/28, F01N3/24, B01D53/34, B01J23/58, F02D41/02

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Jitsuyo Shinan Koho 1926 - 1993

Kokai Jitsuyo Shinan Koho 1971 - 1993

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	JP, A, 2-149715 (Mazda Motor Corp.), June 8, 1990 (08. 06. 90), Lines 3 to 13, column 3 (Family: none)	1-20
A	JP, B2, 1-56816 (Ebara-Infilco Co., Ltd.), December 1, 1989 (01. 12. 89), Lines 3 to 26, column 3 (Family: none)	1-20
A	JP, A, 4-141219 (Mazda Motor Corp.), May 14, 1992 (14. 05. 92), Line 13, column 4 to line 3, column 5 (Family: none)	1-20
A	JP, A, 3-194113 (Mazda Motor Corp.), August 23, 1991 (23. 08. 91), Line 19, column 4 to line 19, column 5 (Family: none)	1-20
A	JP, U, 4-1617 (Toyota Motor Corp.), January 8, 1992 (08. 01. 92), Lines 8 to 19, column 3 (Family: none)	1-20

☒ Further documents are listed in the continuation of Box C.
 ☐ See patent family annex.

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"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

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Date of the actual completion of the international search

July 28, 1993 (28. 07. 93)

Date of mailing of the international search report

August 17, 1993 (17. 08. 93)

Name and mailing address of the ISA/

Japanese Patent Office

Facsimile No.

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INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP93/00777

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	JP, A, 60-182325 (Toyota Motor Corp.), September 17, 1985 (17. 09. 85), Line 13, column 8 to line 7, column 9 & US, A, 4682577	12, 17
Y	JP, A, 53-115687 (Matsushita Electric Ind. Co., Ltd.), October 9, 1978 (09. 10. 78), Line 4, column 1 to line 2, column 2 & JP, B2, 57-27739	19
Y	JP, A, 64-30643 (Matsushita Electric Ind. Co., Ltd.), February 1, 1989 (01. 02. 89), Line 5, column 4 to line 10, column 5 (Family: none)	20

Form PCT/ISA/210 (continuation of second sheet) (July 1992)